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the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20.

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Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided.
1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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SHORT ANSWER Write the answers to the following questions in
the space provided. 22.

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$
- 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O_2 :1 mol H_2O c. What is the mole ratio ...

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Classify each of the following as a homogeneous or heterogeneous substance.
- homogeneous a. sugar homogeneous d. plastic wrap homogeneous b. iron filings heterogeneous e.

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cement sidewalk heterogeneous c. granola bar 2. For each type of investigation, select the most appropriate ...

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