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Chemical kinetics is a term that refers to how fast chemical reactions take place. Some reactions take place very quickly, seemingly instantaneously. Other reactions take place very slowly. Consider for a moment two hypothetical gaseous reactants—A and B—reacting together:

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concept of significant figures has the answer. When multiplying and dividing numbers, you need to round your answers to the correct number of significant figures. To round correctly, follow these simple steps: 1) Count the number of significant figures in each number. 2) Round your answer to the least number of significant figures.

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